

كلية الطب لجان الدفعات

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{ Biochemistry – midterm }

1. What is the mutual property of amylose and cellulose?

- A. α1-4 glycosidic linkage
- B. digestible by humans
- C. consist of reduced glucose residues
- D. They are not branched
- E. None of the above

Answer: D

2. The lipid which doesn't have a phosphate group in its structure is:

- A. Plasmalogens
- B. Ceramide
- C. Glycerophospholipids
- D. Lecithin
- E. Phosphatidylserine

Answer: B

3. The lipid which has 3 conjugated double bonds in its structure is:

- A. Eicosanoids
- B. Thromboxanes
- C. Prostacyclins
- D. Leukotriene

Answer: D

4. The bond between the glycerol and the fatty acids is:

- A. Glycosidic bond
- B. Ether bond
- C. Ester bond
- D. Hydrogen bond

Answer: C

5. The acid with the highest melting point is:

A. palmitic acid

B. Arachidonic acid

C. Stearic acid

D. Oleic acid

E. Linoleic acid

Answer: C

6. Which of the following acids will be 70% unprotonated at pH=4.9

- A. X acid pKa= 5.1
- B. Y acid pKa= 5.9
- C. Z acid pKa= 5.5
- D. V acid pKa= 4.4
- E. N acid pKa= 4.8

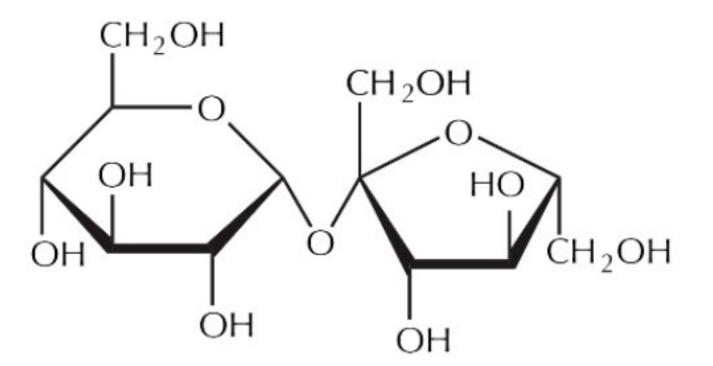
7. How many chiral center(s) is/are present in the ring structure of 2-deoxyribose?

- A. 5
- B. 2
- C. 3
- D. 4

Answer: C

8. The following structure represents:

- A. Sucrose
- B. Maltose
- C. Lactose
- D. Ribose



Answer: A

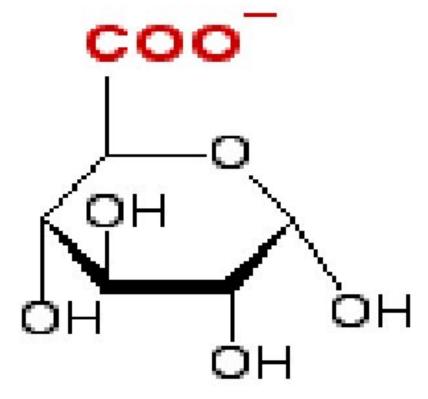
9. Which of the following is a property of Waxes:

- A. Saponifiable
- B. very resistant to rancidity
- C. Butter is an example
- D. Digestible
- E. Tests positive for Acrolein test

Answer: B

10. The following molecule result by?

- A. Strong oxidizing agent
- B. Reducing agent
- C. Weak oxidizing agent
- D. An enzyme



Answer: D

11. The lipoprotein that has the highest lipid content and the lowest protein content is:

- A. HDL
- B. Chylomicrones
- C. LDL

Answer: B

12. Which of the following is an omega 3 fatty acid?

- A. 18:3 Δ9,12,15
- B. 18:3 Δ9,12
- C. 18:1 ^{∆9}
- D. 20: Δ5,8,11,14

Answer: A

13. The following structure describes:

Answer: a conjugate base of a weak acid

14. The pKb of a base is 3.0. what is the pH of a 1 mM solution of the base?

- A. 8
- B. 3
- C. 11
- D. 6

Answer: D

15. Water acquires its high specific heat, boiling point, melting point and other physical properties because:

- A. It is an amphipathic molecule
- B. It has a high ion product of water
- C. It can form hydrogen bonds with each other
- D. It can dissociate to protons and hydroxyl ions
- E. It acts as an amphoteric molecule

Answer: C

16. Protein buffer system:

- A. Acts both intracellularly and extracellularly
- B. Activates carbonic anhydrase when pH is increased
- C. Acts only intracellularly
- D. Is related to the phosphate groups at phosphorylation sites
- E. Has a pKa of 6.1

Answer: A

17. You want to turn a solution containing X moles of Ca(OH)2 into a buffer solution.	
Which of the following should you add?	
A. 2X moles of acetic acid	
B. X/2 moles of acetic acid	
C. 2X moles of HCL	
D. 3X moles of acetic acid	
E. X moles of H2SO4	
	Answer: D
18. Amylopectin is composed of that are connected by and bran	ched at:
A. Galactose residues, alpha-1,4 linkage, carbon number 1	
B. Glucose residues, alpha-1,4 linkage, carbon number 6	
C. Galactose residues, alpha-1,4 linkage, carbon number 6	
D. Ribose residues, alpha-1,4 linkage, carbon number 6	
E. Glucose residues, beta-1,4 linkage, carbon number 1	
	Answer: B
19. The hetero polysaccharides with sulfated sugars, amino sugars and/or	Oxidized
sugars that are mainly derived of glucose and galactose and are found i	n
extracellular matrix are:	
A. Pectin	
B. Dextran	
C. Chitin	
D. Glycosaminoglycans	
E. Cellulose	
	Answer: D
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20. Increasing reabsorption of HCO3- can be the mechanism of compensation of a charge of individual.	ion tor:
A. A starved individual	
B. An asthmatic patient	
C. An uncontrolled diabetic patient	
D. A mountain climber	
E. A controlled diabetic patient	
	Answer: B
21. The glycolipid/s which contain/s a sialic acid:	
A. Gangliosides	
B. Globosides	
C. Cerebrosides	Answer: A