General Anesthetic Agents

- ANESTHESIA: loss of sensation
- General Anesthesia renders the patient:
 - amnesic
 - unconscious while causing muscle relaxation
 - suppression of undesirable reflexes

Theories of anesthesia

- Most anesthetics enhance the activity of inhibitory GABA A-receptors
- Many inhibit activation of excitatory receptors, such as glutamate and nicotinic acetylcholine receptors

Stages of Anesthetic Activity

- Stage I Analgesia
 - conscious but drowsy
 - responses to painful stimuli reduced

Stage II - Excitement

- Iose consciousness
- no longer responds to non-painful stimuli
- responds in a reflex fashion to painful stimuli

Stage III - Surgical anesthesia

- movement ceases and respiration becomes regular
- Stage IV Medullary paralysis
 - respiration and vasomotor control cease
 - death occurs

Properties of Ideal Inhalational Anesthetics

- Rapid, pleasant anesthetic induction and recovery
- Rapid changes in anesthetic depth
- Adequate relaxation of skeletal muscles
- Wide margin of safety
- Absence of toxic effects or other adverse properties in normal doses

Inhalational anesthetics :

Non-halogenated gas:

• Nitrous oxide (NO)

Halogenated hydrocarbons:

- Halothane
- Enflurane
- Isoflurane
- Desflurane
- Sevoflurane
- Methoxyflurane nephrotoxicity.

The important characteristics of Inhalational anesthetics which govern the anesthesia are :

- Solubility in the blood (blood : gas partition coefficient)
- Solubility in the fat (oil : gas partition co-efficient)

Blood : gas partition co-efficient:

- It is a measure of solubility in the blood.
- It determines the rate of induction and recovery of Inhalational anesthetics.
- Lower the blood : gas co-efficient faster the induction and recovery – Nitrous oxide.
- <u>Higher the blood : gas co-efficient</u> slower induction and recovery – Halothane.

Solubility to blood

- Based on blood/gas partition coefficient
- Lowest to highest
- NO
- Isoflurane
- Enflurane
- halothane

POTENCY

- Clinical potency can't be predicted by chemical structure
- Potency correlates w/ lipid solubility
- Defined by MAC

- MAC: concentration of anesthetic gas needed to eliminate movement among 50% of patients challenged by standardized skin incision
- INVERSE of MAC is an index of potency: the more lipid soluble, the LOWER the concentration needed to produce anesthesia
- The lower the MAC, the more potent is the anesthetic

Highest to lowest MAC

- NO
- Ether
- Enflurane
- Isoflurane
- halothane

Individual Inhalation Anesthetics

Nitrous oxide:

- low potency, therefore must be combined with other agents
- rapid induction and recovery
- good analgesic, WEAK anesthetic properties
- can cause diffusion hypoxia
- Safest and least hepatotoxic
- risk of bone marrow depression with prolonged administration.
- laughing gas
- can retard O₂ uptake during recovery, thus 20% of O₂ is always needed

Inhalational anesthetics <u>Halothane:</u>

- It is a potent anesthetic.
- Induction is pleasant.
- It sensitizes the heart to catecholamines.
- It dilates bronchus preferred in asthmatics.
- It inhibits uterine contractions.
- Halothane hepatitis and malignant hyperthermia can occur.

HALOTHANE

- Prototype
- Weak analgesic; POTENT anesthetic
- Usually co-administered w/ N2O, opioids, or local anesthetics
- Metabolized to tissue-toxic hydrocarbons (trifluroethanol) & bromide ion

HALOTHANE

- Disadvantages:
- Reduced myocardial contractility & causes hypotension
- Arrhythmias
- Increase arterial pressure
- Atropine sensitive bradycardia

Inhalational anesthetics

Enflurane:

- Sweet and ethereal odor.
- Generally do not sensitizes the heart to catecholamines.
- Seizures occurs at deeper levels contraindicated in epileptics.
- Caution in renal failure due to fluoride.

Inhalational anesthetics

Isoflurane:

- It is commonly used with oxygen or nitrous oxide.
- It do not sensitize the heart to catecholamines.
- Its pungency can irritate the respiratory system.

ISOFLURANE

- Disadvantages:
 - More pungent odor than halothane
 - Progressive respiratory depression & hypotension

Inhalational anesthetics

Desflurane:

- It is delivered through special vaporizer.
- It is a popular anesthetic for day care surgery.
- Induction and recovery is fast, cognitive and motor impairment are short lived
- Malignant hyperthermia
- It irritates the air passages producing cough and laryngospasm.

METHOXYFLURANE

- potent, high lipid solubility
- Used in child birth
- Disadvantages: Metabolized to flouride; Respiratory and circulatory depression

IV Anesthetics

Thiopental

- barbiturate with very high lipid solubility
- GABA- mimetic
- rapid action because of rapid transfer across blood-brain barrier
- Potent anesthetic, weak analgesic effect
- Little muscle relaxation
- Laryngospasm, not for asthma pt, not w/ porphyria

Thiopental

- It is an ultra short acting barbiturates.
- Consciousness regained within 10-20 mins by redistribution to skeletal muscle.
- It do not increase ICP.
- It is eliminated slowly from the body by metabolism and produce hang over.
- It can be used for rapid control of seizures.

IV anesthetics

Etomidate:

- It is a short acting anesthetic.
- It suppress the production of steroids from the adrenal gland and no repeated injections.
- It is a pro-convulsant and emetic.
- Chorionic villus sampling (CVS) stability is the main advantage over anesthetics.

IV Anesthetic Agents

Ketamine

- analogue of phencyclidine, w/ similar properties (psychotic reactions)
- effect on NMDA-type glutamate receptors
- onset of effect is relatively slow (2-5 minutes)
- produces "dissociative" anesthesia, in which patient may remain conscious and insensitive to pain
- can increase intracranial pressure
- Causes cardiovascular stimulation but not respiratory depression

IV anesthetics

Ketamine:

- A Emergence delirium, hallucinations and involuntary movements occurs in 50% cases during recovery.
- It is useful for burn dressing and trauma surgery
- Dangerous for hypertensive and IHD.

IV anesthetics

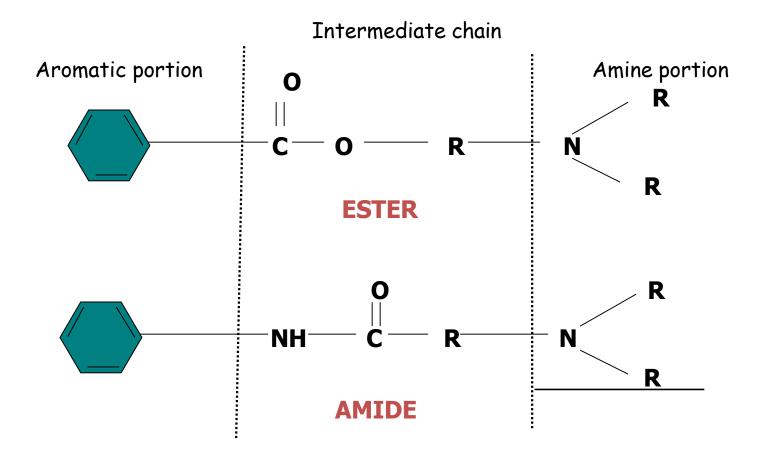
Propofol:

- Most commonly used IV anesthetic.
- Unconsciousness in ~ 45 seconds and lasts ~15 minutes.
- Anti-emetic in action
- Suited for day case surgery residual impairment is less marked.
- LOWERS ICP

Local Anesthetics DESIRABLE CHARACTERISTICS

- Rapid onset of action
- Brief, reversible block of nerve conduction
- Low degree of systemic toxicity***
- Soluble in water and stable in solution
- Effective on all parts of the nervous system, all types of nerve fibers and muscle fibers
- NONE totally meets these optimally yet!!

LAs are Weak Bases



LIPOPHILIC

HYDROPHILIC

Local Anesthetics

Clinical Significance of chemical classification

EPAL

- Biotransformation/duration of action
 - ESTERS are rapidly metabolized in the plasma by a cholinesterase
 - AMIDES are more slowly destroyed by liver microsomal P450 enzymes.

Local Anesthetics MECHANISM OF ACTION

- LA have greatest affinity for sodium channel in inactivated state and slows its reversion to the resting state.
- Administration:
- Topical, injected into nerves, epidural or subarachnoid space (spinal)
- Notes:
- Reduced pH, as in inflamed tissues reduces effectiveness (cationic form predominates)
- Co-administered with vasoconstrictor epinephrine (1:100,000)

Local anesthetics (LAs)

The esters

The esters

- Benzoic acid derivatives
- P-aminobenzoic acid derivatives
- Acetanilide-lidocaine
- Toluidine prilocaine
- thiophene ultracaine

Benzoic Acid Derivatives

- Cocaine
- Cyclomethicaine
- Hexylcaine
- Isobucaine
- Meprylcaine
- Piperocaine

P-aminobenzoic acid Derivatives

- Benzocaine
- Buta<u>caine</u>
- Propoxy<u>caine</u>
- Processine
- Propara<u>caine</u>
- Chlorprocaine
- Tetracaine

Cocaine

- medical use limited to surface or topical anesthesia (corneal or nasopharyngeal)
- avoid epinephrine because cocaine already has vasoconstrictor properties (EXCEPTION!!!)
- A toxic action on heart may induce rapid and lethal cardiac failure
- A marked pyrexia is associated with cocaine overdose

Benzocaine

- Used to produce anesthesia of mucous membranes and to suppress gag reflex during endoscopy
- methemoglobinemia

Procaine

- Used for *infiltration* because of low potency and short duration but most commonly used for **spinal** anesthesia
- Short local duration -produces significant vasodilation. Epinephrine used to prolong effect
- DOC for patients with liver disease

Tetracaine

- topical, infiltration and spinal anesthesia
- Frequently used for topical ophthalomogical anesthesia
- slow onset and more prolonged effect than procaine (longest duration of the esters)

Local anesthetics (LAs)

The AMIDES

The amides

- Lidocaine
- Mepivacaine
- Bupiva<u>caine</u>
- Prilocaine
- Etidocaine
- Ropiva<u>caine</u>
- Levobupiva<u>caine</u>
- Articaine

LIDOCAINE:

- most widely used LA
- effective by all routes
- faster onset, more intense, longer lasting, than procaine
- good alternative for those allergic to ester type
- more potent than procaine but about equal toxicity
- more sedative than others

Mepivicaine:

- effective by all routes except topical
- similar onset and duration as lidocaine
- <u>more toxic to neonates</u> so not used in obstetrical anesthesia
- fetus poorly metabolizes mepivicaine

Bupivacaine:

- no topical effect
- slower onset and one of longer duration agents
- unique property of sensory and motor dissociation
- can provide sensory analgesia with minimal motor block
- has been popular drug for analgesia during labor

Ropivacaine:

- enantiomer of bupivacaine (S stereoisomer)
- no topical effectiveness
- clinically ~ equivalent to bupivacaine
- similar sensory versus motor selectivity as bupivacaine
- significantly less cardiotoxicity than bupivacaine at equipotent doses.
- Long-acting LAs are very potent and may cause cardiac arrest with a misplaced injection or relative overdose

THANK YOU

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